

MINIMIZING PFC AND HAP EMISSIONS DURING ULTIMA HDP-CVD PROCESSING: PSG, USG, AND FSG FILMS

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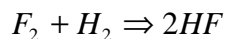
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In support of a voluntary commitment to minimize process emissions, considerable effort has been expended by semiconductor manufacturers, OEMs, and materials suppliers to identify, quantify, and reduce emissions for individual process steps. This paper describes an emissions reduction program for three Ultima HDP-CVD processes: PSG, USG, and FSG deposition and chamber clean. Quantitative QMS and FTIR measurements show that during thin film deposition the SiH₄ is fully utilized, while for FSG films, the SiF₄ emissions associated with the deposition step is 7 mg/μm. The NF₃ remote microwave process was shown to effectively eliminate PFC emissions from Ultima HDP-CVD cleaning (3 mg/μm). For all three processes, the NF₃ utilization is better than 99 %. The MMTCE for this process is 0.15×10^{-9} , which represents a reduction of more than 90 %. Byproducts of the remote NF₃ clean process were identified as unreacted NF₃, SiF₄, F₂, and HF: 92-100 % of the influent fluorine is accounted for. A Delatech CDO was found to be an effective solution for abating the fluorine effluent from Ultima HDP-CVD processes. Indeed, the Delatech CDO removes 99.8 % of the influent fluorine during Ultima HDP-CVD processing.

INTRODUCTION

Semiconductor manufacturers are concerned about atmospheric emissions associated with wafer fabrication. In support of a voluntary commitment to minimize process emissions, considerable effort has been expended by semiconductor manufacturers, OEMs, and materials suppliers to identify, quantify, and reduce emissions for individual process steps. Initially, industry concern focused upon perfluorocompound (PFC) emissions because they are suspected of contributing to anthropogenic climate change. PFCs in the atmosphere are believed to originate exclusively from anthropogenic uses and, because of their long lifetimes and strong infrared absorbances, could contribute to global warming. The predominant use of PFCs in semiconductor manufacturing is to clean process chambers following thin film CVD. As part of the semiconductor industry's voluntary commitment to minimize PFC emissions, Applied Materials developed a new process for cleaning CVD tools: NF₃ remote microwave clean. This paper describes the characterization of PFC and HAP emissions during Ultima HDP-CVD thin film deposition and chamber cleaning.

Process emissions were identified and quantified by quadrupole mass spectrometry (QMS) and fourier transform infrared spectroscopy (FTIR) measurements (Section 2). Three Ultima HDP-CVD processes were characterized (Section 3): phosphosilicate glass (PSG), undoped silicate glass (USG), and fluorine doped silicate glass (FSG). Besides NF_3 , the other gaseous emissions are HF, F_2 , and SiF_4 . Although NF_3 emissions were dramatically reduced, this characterization determined that F_2 emissions during the clean process were substantial. While F_2 has no global warming potential, it is hazardous material that should not be released to the atmosphere. In section 4, F_2 emissions using a Delatech CDO are described. The Delatech CDO uses a two stage process. The F_2 byproducts are first reduced with H_2 in a combustion chamber:



The HF byproduct is then scrubbed as HF(aq) using a neutral water scrubber and the waste water that is generated subsequently treated for disposal.

2. EXPERIMENTAL METHODOLOGY

PSG (8,000 Å), USG (15,000 Å), and FSG (25,000 Å) films were deposited in a 200 mm Applied Materials' Ultima HDP-CVD chamber. The deposition gases are SiH_4 , PH_3 , and SiF_4 . O_2 and Ar were the other process gases. After processing 5 PSG wafers, a periodic remote microwave NF_3 process was used to clean the chamber. For the USG and FSG depositions, the process chamber was cleaned after each process.

Process byproducts were measured by QMS and FTIR downstream of the process pump and abatement system. The process was sampled through a 1/4 inch VCR compression fitting welded to the exhaust elbow of the process pump. The gases of interest are therefore diluted by any pump purge (section 2.3). The sample inlet pressure was 750 torr throughout as measured by a capacitance manometer. Sample lines were 1/8" inch stainless steel GC tubing, except the 3 ft. length of 1/4 inch EP 316L tubing connecting the QMS and FTIR instruments. All sample lines were heat traced to $\sim 100^\circ\text{C}$. Before monitoring the Ultima HDP-CVD process, all sample lines were fluorine passivated by running the remote NF_3 clean process in a clean chamber.

2.1 QMS Measurements of NF_3 , SiF_4 , and F_2

The processes were monitored with a UTI Qualitrac QMS having a 200 amu mass filter. The sample inlet was differentially pumped for fast sampling of a slightly subambient atmosphere i.e. the QMS response time must be much shorter than changes in the gas composition of each process step. The mass locations were determined using N_2 (7, 14, 28 amu) and NF_3 (71 and 52 amu). The QMS instrument was directly calibrated for NF_3 , SiF_4 , and F_2 using 1 % gas standards. Five (5) point or better calibration curves were measured using dynamic dilution methods to generate concentrations $< 10,000$ ppm. Due to sensitivity changes, the QMS analyzer was calibrated each day.

2.2 FTIR Measurements of SiH₄, PH₃, and HF

SiH₄, PH₃, and HF concentrations were determined using a Midac I2000 FTIR spectrometer with a HgCdTe detector and a Axiom gas cell (0.1 m or 0.01 m). Absorbance spectra were collected at 1.0 cm⁻¹ resolution (8 scans). The FTIR spectrometer was directly calibrated *ex situ* for HF measurement using a 0.5 % HF gas standard. These concentrations were determined from a single point calibration (540 ppm.m or 55 ppm.m reference spectra). Absorbance from the rotational-vibrational transition at 4038 cm⁻¹ was used as a monitor of the HF concentration. The SiH₄ calibration was obtained using the tool gas panel by diluting the SiH₄ flow (10 -75 sccm) with the pump purge (section 2.3). The Si-H stretch region (2174-2199 cm⁻¹) was used for SiH₄ measurements. A single point calibration was used for SiH₄ determinations (34.4 ppm.m) since the calibration is linear over the concentration range. A library spectrum was used for PH₃ concentration measurements (100 ppm.m) where the region 2302-2381 cm⁻¹ was used for concentration measurements.

2.3 Pump Purge Measurement

Because measurements are made downstream of the process pump, the process byproducts are diluted by the pump purge. It is important to know the total gas flow to determine volumetric emissions i.e. standard cubic centimeters (scc). The pump purge was determined by flowing NF₃ from the gas panel and measuring its concentration using the QMS which has been independently calibrated for NF₃ (section 2.1). The dilution factor is obtained from the slope of NF₃ concentration versus NF₃ gas flow. For measurements downstream of the abatement system, NF₃ cannot be used to determine the total gas flow since it thermally decomposes above 500 °C. The total pump purge and makeup air flowing through the Delatech CDO was estimated by flowing He (60 sccm) using the gas panel on the process tool. The He concentration (340 ppm) downstream of the Delatech CDO was measured using the mass spectrometer which had been independently calibrated for He measurements. The total purge causing this dilution is calculated to be 177 slm.

3. ULTIMA HDP-CVD PROCESSING

3.1 PSG Deposition

The 4250 cm⁻¹ absorbance spectrum during PSG film deposition is shown in Fig. 1. No SiH₄ or PH₃ features are observed in the region 2000 cm⁻¹ to 2500 cm⁻¹. The peaks at 835 cm⁻¹ and 1235 cm⁻¹ are due to PSG film deposition on the windows of the FTIR gas cell (i.e. they are observed throughout the effluent characterization). Figure 2 shows the concentration of SiH₄ and PH₃ (FTIR measurement) during Ultima HDP-CVD PSG processing. These time profiles confirm that the SiH₄ and PH₃ are completely utilized by this process (SiH₄ and PH₃ are below the FTIR detection limits).

The 4250 cm⁻¹ absorbance spectrum (Fig. 3) and 200 amu mass spectrum (Fig. 4) collected during the NF₃ clean process identify the by-products as NF₃, SiF₄, PF₅, F₂, and HF. We conclude that the phosphorous is removed as PF₅ from its QMS cracking pattern: 107 amu (PF₄), 88 amu (PF₃), 69 amu (PF₂), and 50 amu (PF). The concentrations of NF₃, SiF₄, and F₂ during Ultima HDP-CVD PSG processing are shown in Figs. 5. The HF concentration, due to hydrogen in the PSG film, measured by FTIR during wafer processing is shown in Fig. 6. The first two processes (PASS 1 and PASS 2) were run with a clean chamber (i.e. no dep.) for the purpose of fluorine passivating the gas lines (Section 2). The NF₃ concentration remains low (<1000 ppm) due to a high utilization (99 %). Once the MW power is applied, there is a sharp increase in the SiF₄ concentration as the chamber is cleaned. The dominant byproduct of the clean process is elemental F₂.

| PSG CLEAN | | | | | | | AVE | |
|------------------------|--------------|--------------|----------|----------|----------|----------|--------------|---------------|
| | PASS1 | PASS2 | 1 | 2 | 3 | 4 | (scc) | (g/um) |
| <i>NF3(scc)</i> | 60 | 41 | 44 | 39 | 38 | 37 | 39 | 0.03 |
| <i>SiF4(scc)</i> | 73 | 60 | 687 | 672 | 662 | 651 | 668 | 0.78 |
| <i>PF5</i> | 4 | 3 | 34 | 34 | 33 | 33 | 34 | 0.5 |
| <i>F2(scc)</i> | 8153 | 9668 | 7707 | 8670 | 8908 | 8940 | 8556 | 3.63 |
| <i>HF(scc)</i> | 122 | 189 | 547 | 539 | 520 | 510 | 529 | 0.12 |
| | | | | | | | | |
| <i>Utilization (%)</i> | 99 | 99 | 99 | 99 | 99 | 99 | 99 | |
| <i>F-Balance</i> | 0.79 | 0.93 | 0.89 | 0.98 | 0.99 | 1.00 | 0.96 | |
| | | | | | | | | |
| PSG DEPOSITION | | | | | | | AVE | |
| | PASS1 | PASS2 | 1 | 2 | 3 | 4 | (scc) | (g/um) |
| <i>SiH4(scc)</i> | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| <i>PH3(scc)</i> | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |

Table I.

Integrating under the concentration profiles (Figs. 5 and 6) allows the volumetric emissions to be calculated. The corresponding SiH₄, PH₃, NF₃, SiF₄, PF₅, F₂, and HF emissions for the Ultima HDP-CVD PSG process are summarized in Table I. Note that the volumetric emission (scc) represents the cumulative emission for the 5 wafer periodic clean, whereas, the mass emission (g/μm) has been normalized for the film thickness (8,000 Å) and number of wafers processed per clean (5). Rows 3-7 summarize the NF₃, SiF₄, PF₅, F₂, and HF emissions, respectively. Using the volumetric emissions of NF₃, and ratioing this to the NF₃ influent, the NF₃ utilization is calculated to be 99 % (row 9). Using the volumetric emissions of the fluorine byproducts, and their molecular formulae, the total amount of fluorine emitted can be calculated. Comparing this value to the fluorine influent gives the fluorine balance. The fluorine influent was obtained by multiplying the NF₃ flow rate by the clean time. For the 25 processed wafers, the fluorine balance is 96 % indicating that all of the fluorine byproducts have been accounted for.

3.2 USG Deposition

Concentrations of SiH₄, HF, NF₃, SiF₄, and F₂ are shown in Figs. 7 and 8. As for PSG deposition, the SiH₄ is completely utilized during USG HDP-CVD (Fig. 7). Volumetric emissions during Ultima HDP-CVD USG processing are summarized in Table II. The NF₃ utilization for this clean process is 100 % with 92 % of the fluorine being accounted for.

| | 1 | 2 | 3 | AVE |
|-----------------------|-------------|-------------|-------------|-------------|
| <i>NF3 (scc)</i> | 30 | 18 | 18 | 22 |
| <i>SiF4 (scc)</i> | 270 | 221 | 255 | 249 |
| <i>F2 (scc)</i> | 6939 | 7054 | 7175 | 7056 |
| <i>HF (scc)</i> | 18 | 18 | 18 | 18 |
| | | | | |
| <i>F Balance</i> | <u>0.91</u> | <u>0.91</u> | <u>0.94</u> | 0.92 |
| <i>Utilization(%)</i> | <u>0.99</u> | <u>1.00</u> | <u>1.00</u> | 1.00 |

Table II.

3.3 FSG Deposition

Concentrations of SiH₄, HF, NF₃, SiF₄, and F₂ during the deposition and clean process are shown in Figs. 9 and 10. The SiH₄ is completely utilized during FSG HDP-CVD (Fig. 9). Note that these FTIR measurements cannot be used to establish SiF₄ emissions during the clean due to the high concentration (Fig. 10) saturating the infrared region. FTIR data are, however, used to determine the SiF₄ emissions during the deposition. Volumetric emissions during Ultima HDP-CVD FSG are summarized in Table III and IV for the deposition and clean processes, respectively. The NF₃ utilization for this clean process is 99.7 %, and 100 % of the fluorine is accounted for. The NF₃ influent value was obtained by running the clean process with no microwave power applied. This method accounts for any variation in the MFC calibration.

| <i>DEP (scc)</i> | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 |
|------------------------|----------|----------|----------|----------|----------|----------|----------|----------|----------|-----------|-----------|-----------|-----------|
| <i>SiF₄</i> | 37 | 29 | 35 | 39 | 29 | | 30 | 28 | 38 | 55 | 34 | 34 | 53 |
| <i>HF</i> | 62 | 54 | 53 | 57 | 50 | | 53 | 50 | 54 | 56 | 47 | 49 | 58 |
| <i>SiH₄</i> | 4 | 0 | 0 | 3 | 0 | | 0 | 0 | 0 | 0 | 0 | 0 | 1 |

Table III

| | 1 | 2 | 3 | AVE |
|------------------|----------|----------|----------|------------|
| <i>NF3 (scc)</i> | 40 | 27 | 29 | 32 |

| | | | | |
|------------------------------|-------------|-------------|-------------|--------------|
| <i>SiF₄ (scc)</i> | 793 | 849 | 817 | 820 |
| <i>F₂ (scc)</i> | 13559 | 13443 | 13562 | 13522 |
| <i>HF (scc)</i> | 99 | 99 | 99 | 99 |
| | | | | |
| <i>F Balance</i> | <u>1.00</u> | <u>1.00</u> | <u>1.00</u> | 1.00 |
| <i>Utilization (%)</i> | <u>99.6</u> | <u>99.7</u> | <u>99.7</u> | 99.7 |

Table IV

4. ULTIMA HDP-CVD PROCESS ABATEMENT

The NF₃ remote plasma source has been shown (Section 3) to essentially eliminate PFC emissions from Ultima HDP-CVD processes. The dominant byproduct during the clean is now elemental F₂. Abatement of both USG (15 kA) and FSG (25 kA) processes by a Delatech CDO was characterized. For each process, three wafers were processed. The concentrations of NF₃, SiF₄, and F₂ during CVD and the subsequent clean are shown in Fig. 11. At the start of the clean process, the NF₃ concentration spikes to ~90 ppm. During the clean and overetch, however, the NF₃ is at baseline levels. The SiF₄ concentration increases to 5 ppm during the main etch then returns to baseline levels once the chamber is clean. The NF₃ utilization is >99 %, in agreement with the earlier studies. The dominant byproduct of the remote NF₃ clean process downstream from the reactor is F₂, whose concentration is ~20 ppm throughout the clean process. The volumetric NF₃, SiF₄, and F₂ emissions downstream of the Delatech CDO are summarized in tables V for the USG and FSG processes. These emissions are based upon a purge flow of 177 slm (section 2.3). Also shown in table V are the process emissions before the Delatech CDO. Elemental F₂ emissions for both the USG and FSG process are reduced by 99.8 % downstream of the Delatech CDO.

| | USG (15 kA) | | FSG (25 kA) | |
|------------------------------|-------------|-----------|-------------|-----------|
| | BEFORE CDO | AFTER CDO | BEFORE CDO | AFTER CDO |
| NF₃ (scc) | 22 | 7 | 32 | 10 |
| SiF₄ (scc) | 249 | 1 | 820 | 2 |
| F₂ (scc) | 7,056 | 13 | 13,522 | 31 |
| HF (scc) | 100 | n/a | 100 | n/a |

Table V. Volumetric emissions during USG and FSG processing.

5. SUMMARY

Emissions from three Ultima HDP-CVD processes have been determined by quantitative QMS and FTIR measurements. During PSG, USG, and FSG deposition the SiH_4 is fully utilized ($0 \text{ mg}/\mu\text{m}$). For FSG films, the SiF_4 emission associated with the deposition step is $7 \text{ mg}/\mu\text{m}$. The NF_3 remote microwave process was shown to effectively eliminate PFC emissions from Ultima HDP-CVD cleaning ($3 \text{ mg}/\mu\text{m}$). The MMTCE for this process is 0.15×10^{-9} , which represents a reduction of more than 90 % over current practice. Byproducts of the remote NF_3 clean process were identified as SiF_4 , F_2 , HF , and unreacted NF_3 . Most concern focused on the elemental F_2 generated. A Delatech CDO was found to be very effective at abating the fluorine effluent from the Ultima HDP-CVD processes. The Delatech CDO removes 99.8 % of the influent fluorine.

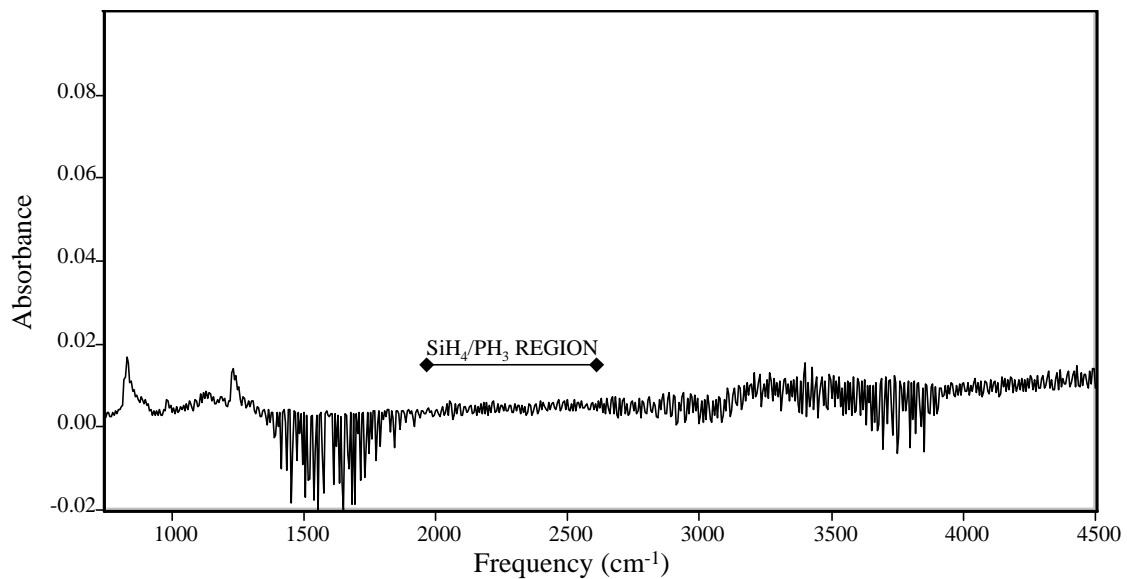


Figure 1.

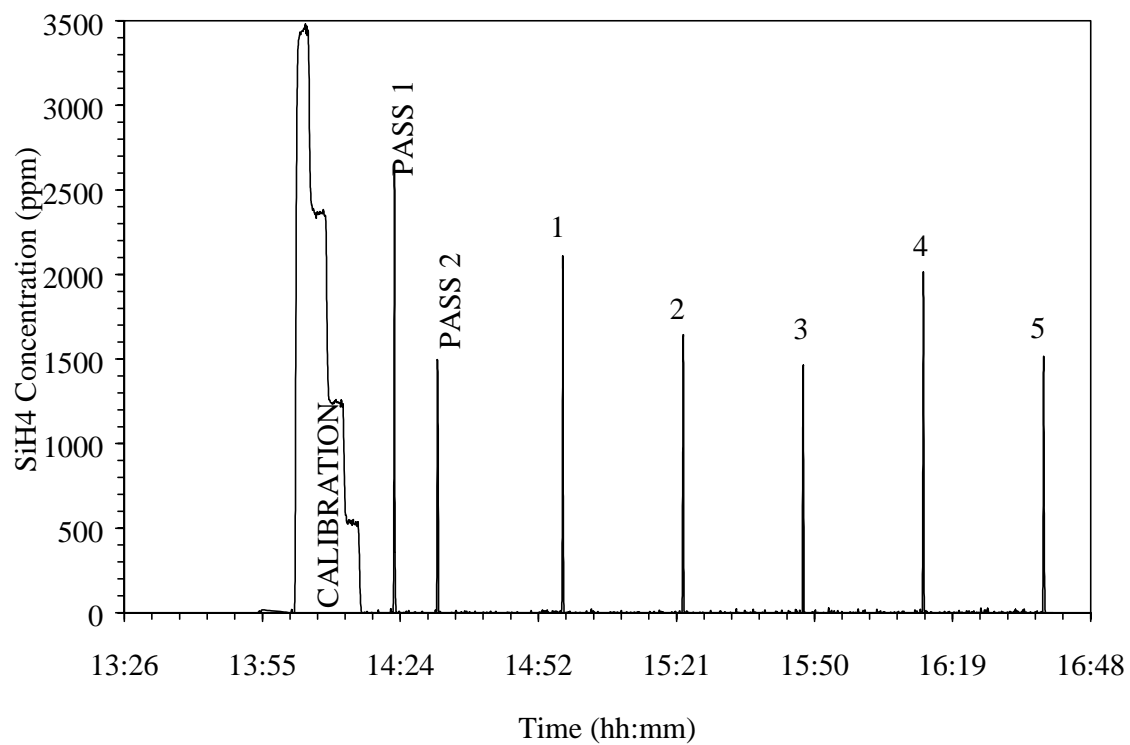


Figure 2.

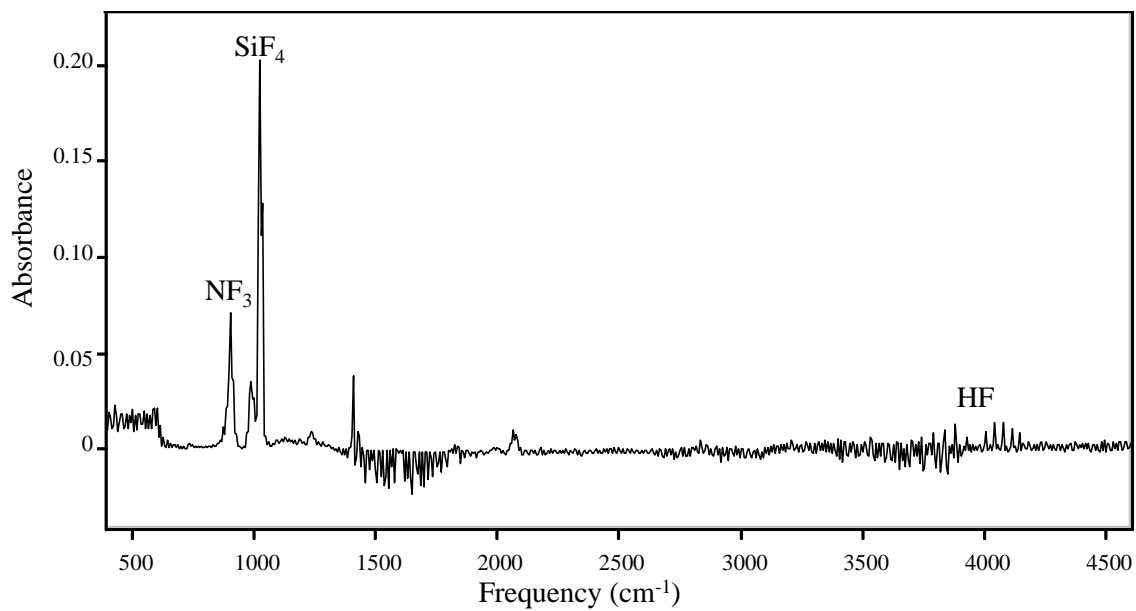


Figure 3.

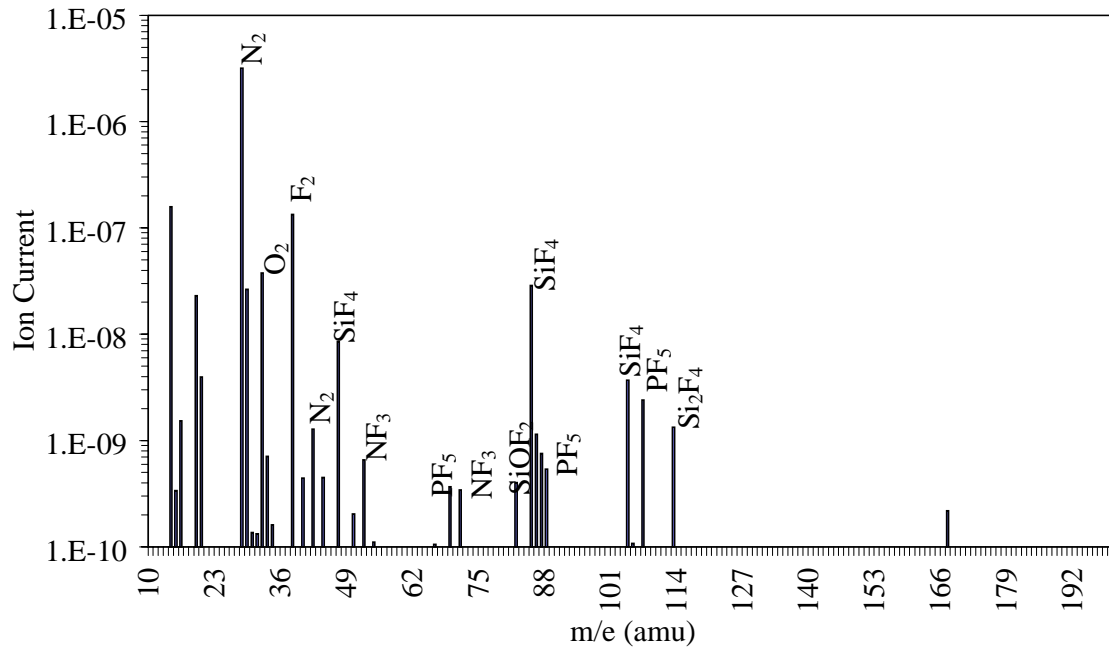


Figure 4.

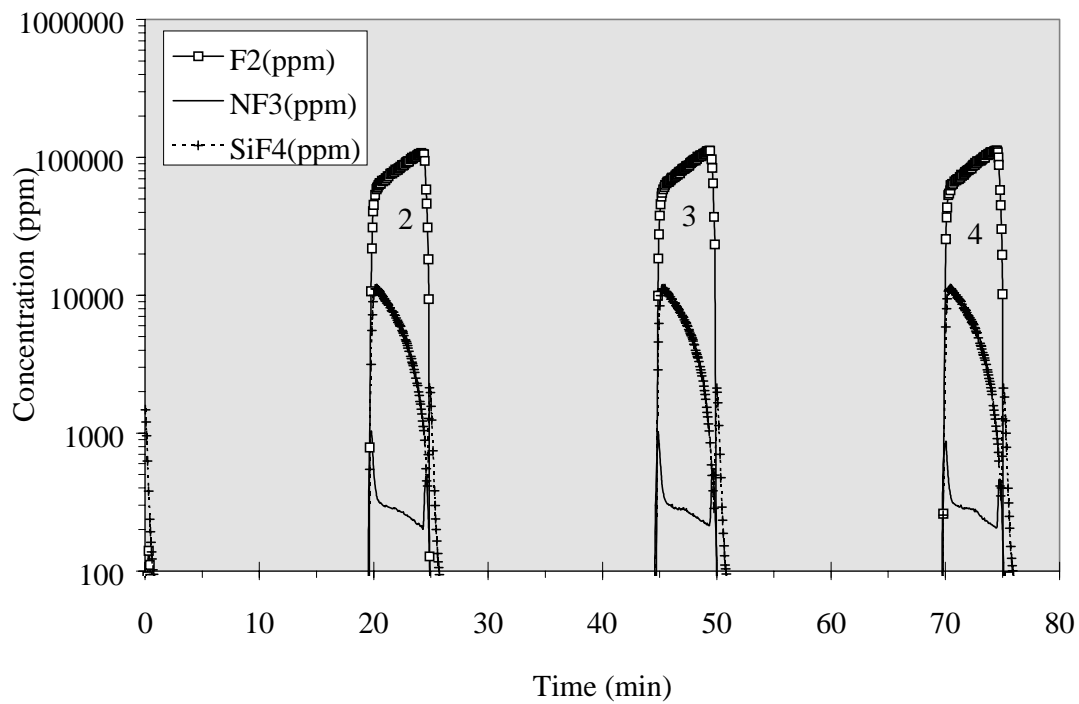


Figure 5.

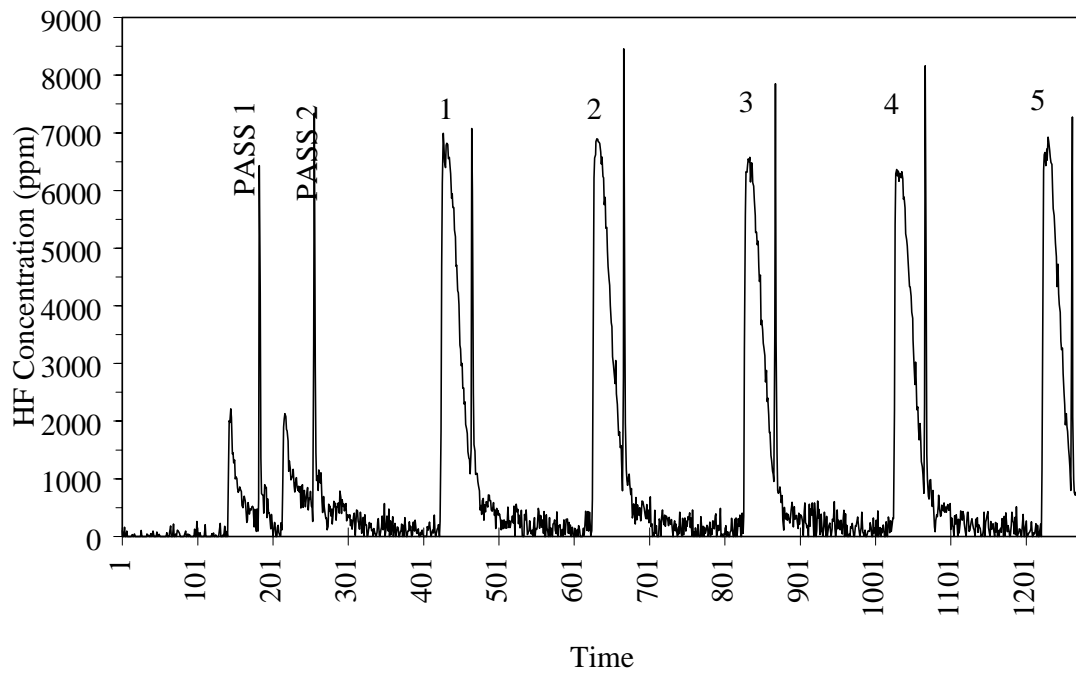


Figure 6.

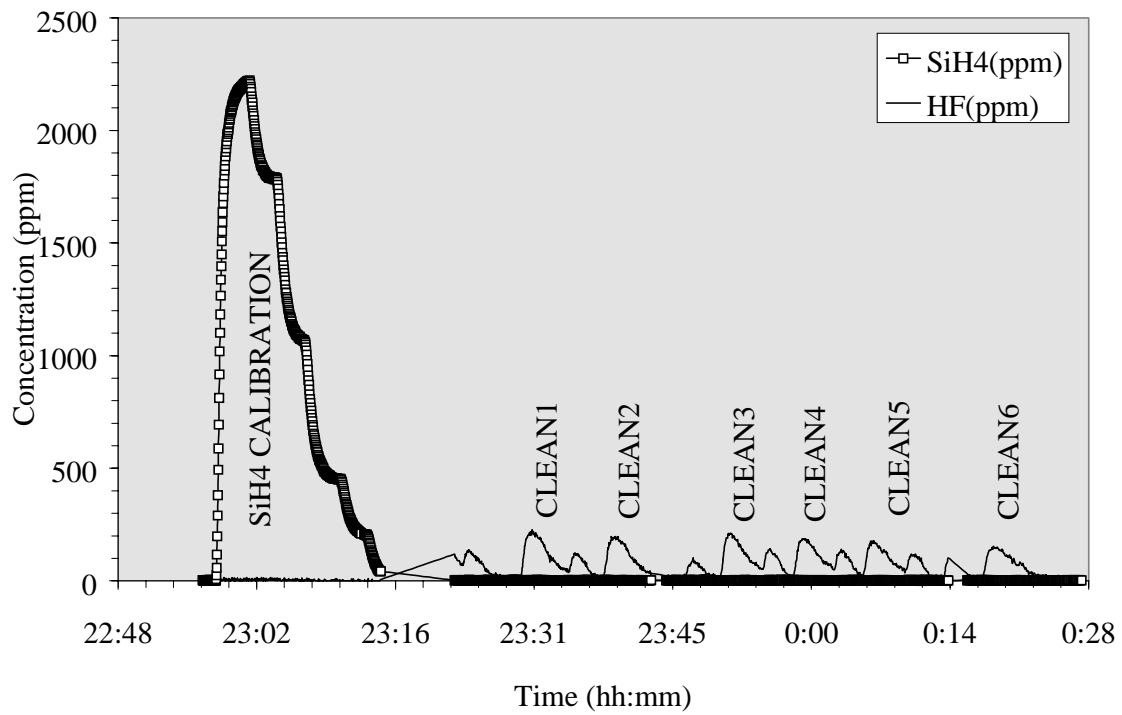


Figure 7.

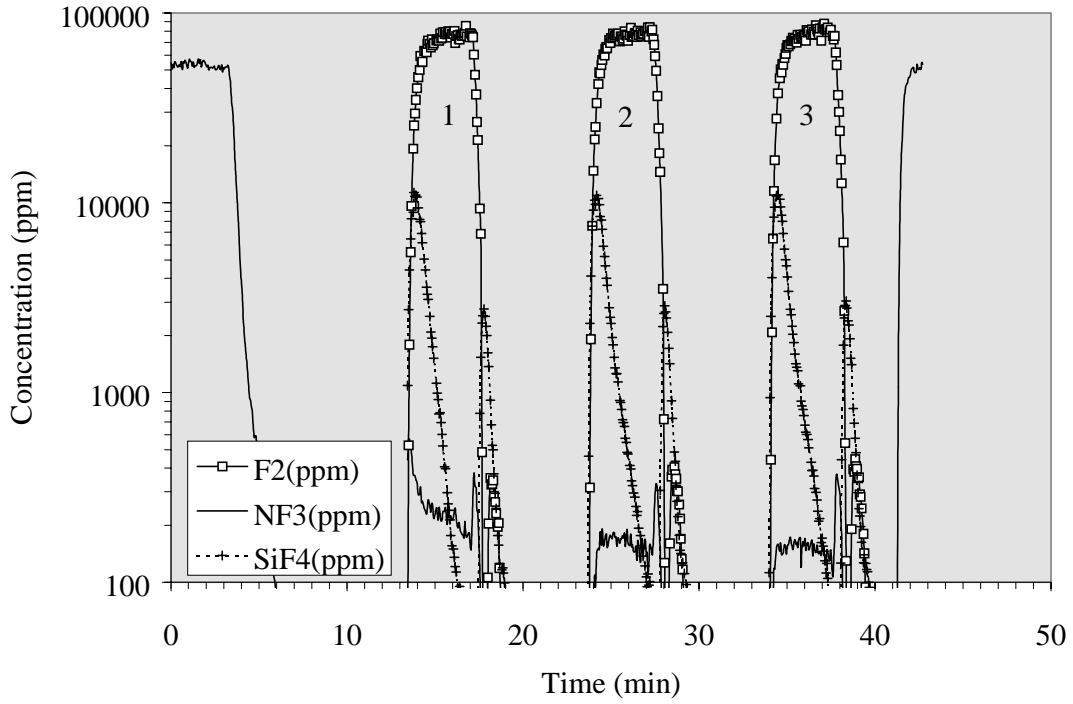


Figure 8.

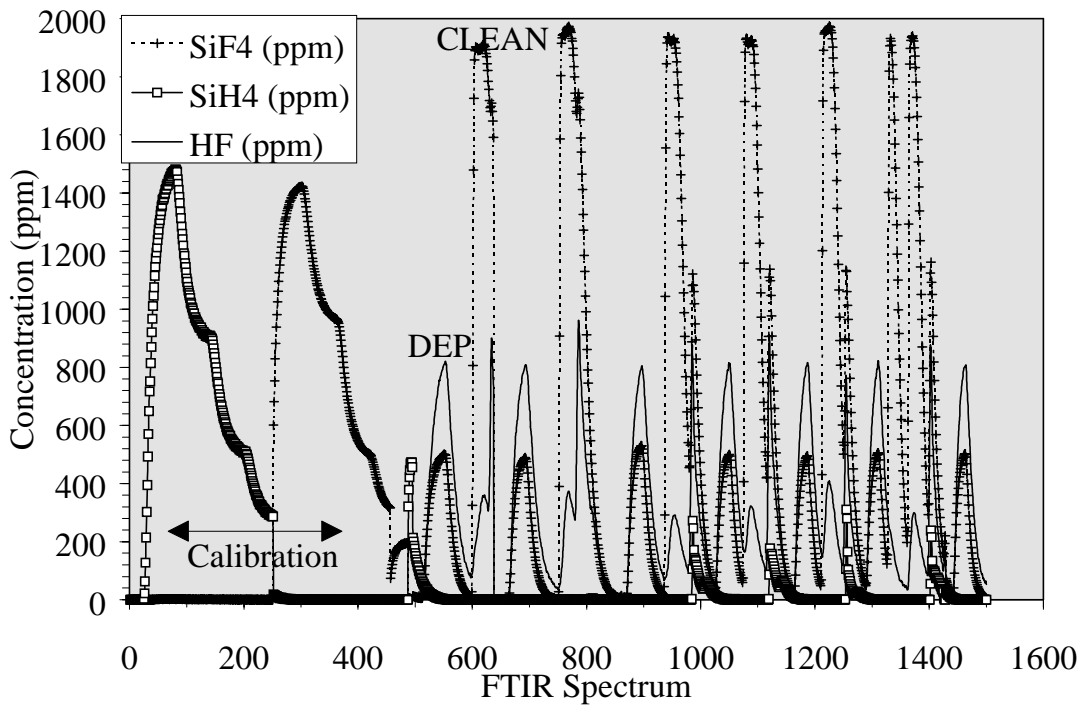


Figure 9.

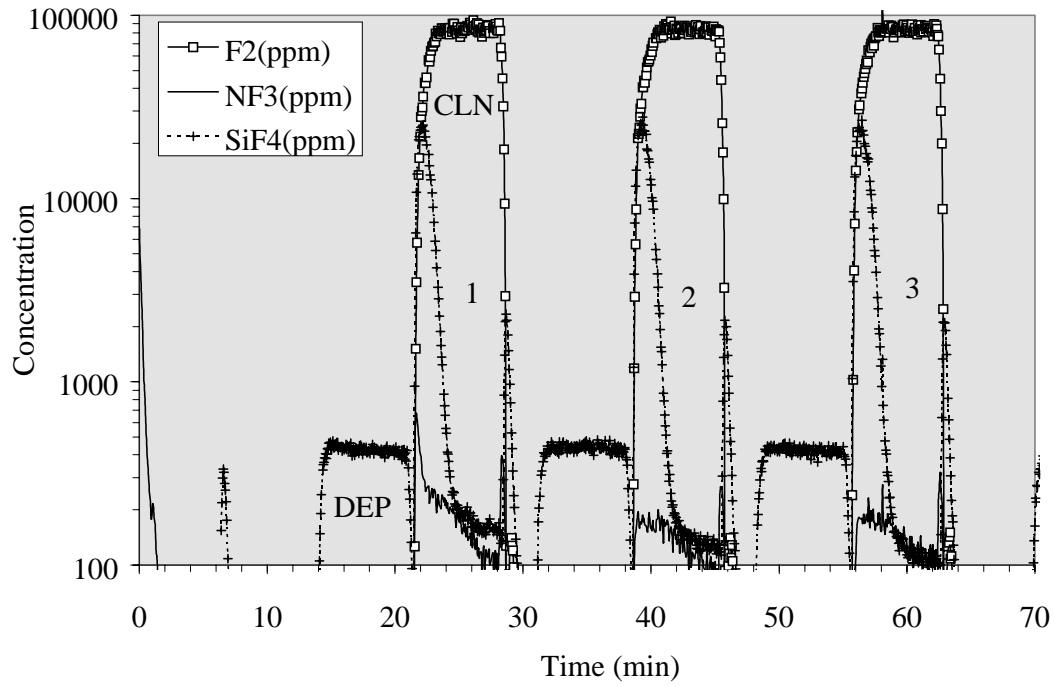


Figure 10.

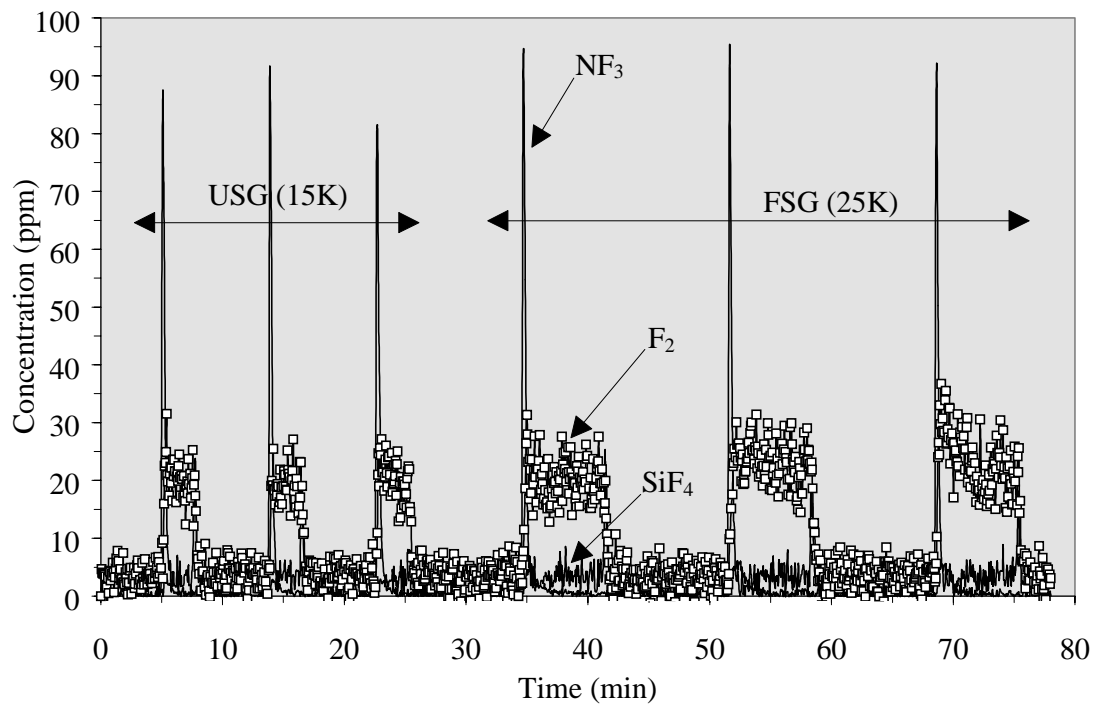


Figure 11.